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10 **Increasing the reaction kinetics of alkali activated fly ash binders for**  
11 **stabilisation of a silty sand pavement sub-base**

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35

36    **Abstract**

37

38    The paper addresses several options to improve the reaction kinetics of alkali activated low  
39    calcium fly ash binders for soil stabilisation in road platforms. For that purpose, an experimental  
40    program was established to assess the strength evolution, with time, of different binders, based  
41    on ash, lime, sodium chloride and alkali solutions, applied in the stabilisation of a silty sand.  
42    The tests included unconfined compression strength tests, triaxial tests and seismic wave  
43    measurements performed at different curing periods. The results were compared with a binder  
44    made of Portland cement and a commercial additive specifically designed for soil stabilization  
45    in road applications. The activated ash mixtures with lime were the most performing producing  
46    a significant increase in the reactions development and, consequently, in the strength gain rate.  
47    The sodium chloride significantly improved the lime and lime-ash mixtures, but provided only  
48    a slight improvement in the activated ash mixtures.

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51    Keywords: soil stabilization; fly ash; lime; alkaline activation, Portland cement

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54     **Introduction**

55     With the worldwide increase in constructed surface area, the need for local soil stabilization  
56     or improvement keeps growing. Different methods have been used for decades, and several of  
57     those are still being improved and refined, while new ones keep arising as technology  
58     progresses. According to Mitchell (1981) and van Impe (1989) the stabilisation /  
59     improvement techniques can be divided in the following groups: consolidation; grouting;  
60     mechanical, chemical and physical stabilisation; and soil reinforcement. Almost all of the  
61     mentioned groups comprise one or more techniques that can be very efficiently applied in  
62     transport infrastructures. Compaction, consolidation, grouting and soil reinforcement can be  
63     used to improve the bearing capacity and deformability properties of embankments or slopes;  
64     while mechanical (surface compaction) and chemical stabilisation are very common when  
65     dealing with superficial layers.

66                 In particular, chemical stabilisation is very efficient when the soils available nearby  
67     cannot present an adequate performance based only on mechanical stabilisation. The use of a  
68     soil-binder combination specifically developed for subgrade layers can significantly increase  
69     its strength and stiffness, avoiding the extraction of more performing materials from distant  
70     locations (Ingles & Metcalf, 1972; Little, 1995; Sherwood, 1993; Little & Nair, 2009).

71                 The most common binders for surface stabilisation are based on Ordinary Portland  
72     cement (OPC) and/or lime (Petry & Little, 2002; Xing et al, 2009). Due to the pozzolanic  
73     properties of common clay minerals, the high amount of calcium present in both slaked and  
74     unslaked lime makes it a very effective option to decrease plasticity and increase workability,  
75     with also some significant strength and stiffness increase if dry weight percentages above 6%  
76     are used (Cristelo et al, 2009; Little et al, 2010; Al-Mukhtar et al, 2014; Gullu, 2015; Han &  
77     Cheng, 2015; Elkady, 2016). When dealing with silty to sandy soils, OPC is usually the most  
78     interesting option (Consoli et al, 2011a; Bahar et al, 2004; Goodary et al, 2012; Rios et al.,  
79     2012, 2014). However, if a significant increase in strength is necessary in a clayey soil, the

80 most consensual strategy is probably to mix the soil with lime, before mixing the resulting  
81 granular material with OPC (Horpibulsuk et al., 2006; Yusuf et al, 2001; Yong & Ouhadi,  
82 2007; Oza & Gundaliya, 2013; Khemissa and Mahamedi, 2014; Saride et al, 2013). Several  
83 binders have been tested, with more or less success (Kolias et al, 2005; Basha et al 2005;  
84 Kaniraj & Havanagi, 1999; Kamei et al, 2013; Yilmaz & Ozaydin, 2013; Gurbuz et al, 2015),  
85 but in almost every case OPC plays a major role in the stabilisation process.

86 Based on the literature review, it becomes clear that cement is the most important  
87 constituent in chemical stabilisation of soil subbases, producing significant strength and  
88 stiffness increase with little or no setbacks whatsoever. Nevertheless, the decreasing tolerance  
89 regarding the environmental concern has the production of clinker as a major target, mostly  
90 due to the high amount of CO<sub>2</sub> released. A value between 5% and 8% of the overall CO<sub>2</sub>  
91 released to the atmosphere is estimated to be originated by OPC production (Provis & van  
92 Deventer, 2014; Scrivener & Kirkpatrick, 2008). Therefore, there is an ongoing research  
93 effort targeting the development of more sustainable binders (Juenger et al., 2011). In  
94 particular, the use of waste materials is highly encouraged, since it allows an increase in  
95 resource efficiency, while contributing also to enhancing the circular economy (by reducing  
96 landfilled waste).

97 The alkaline activation technique is particularly adequate to create binders based on  
98 residues, such as fly ash or ground granulated blast furnace slag, which constitute very  
99 effective options due to their amorphous aluminosilicate microstructure. It consists on a  
100 reaction between aluminosilicate materials and alkali or alkali-based earth substances, such as  
101 sodium (Na) or potassium (K), or an alkaline earth ion, such as calcium (Ca). The reactions  
102 can be summarized in the following sequence. First, there is the destruction, by the high  
103 hydroxyl (OH-) concentration in the alkaline medium, of the Si-O-Si, Al-O-Al and Al-O-Si  
104 covalent bonds present in the vitreous phase of the original semi-amorphous aluminosilicate  
105 (i.e. the precursor). The Si and Al ions are released into the solution as they become available

106 and; at the same time, the alkaline cations – usually Na<sup>+</sup> or K<sup>+</sup>, depending on the activator –  
107 compensate the excess negative charges associated with the modification of the aluminium  
108 coordination during the dissolution phase. The resulting products accumulate for a period of  
109 time, forming an ion-rich solution, which finally precipitates and reorganizes into more stable  
110 and ordered Si-O-Al and Si-O-Si structures. If calcium is predominant, relatively to the  
111 sodium or potassium, the dissolved Al-Si ions will diffuse from any solid surface, which  
112 favours the production of a C-S-H gel phase. Otherwise, the Si and Al ions will be able to  
113 accumulate around the nuclei points, sharing all the oxygen ions and forming a Si-O-Al and  
114 Si-O-Si three-dimensional structure (the formation of Al-O-Al is not favoured). The resulting  
115 product is an amorphous alumina-silicate gel, which evolves, with curing time and  
116 crystallization, from an Al-rich phase to a Si-rich phase. The crystallization, starting almost  
117 immediately after the precipitation, is responsible for the hardening of the gel, which  
118 eventually matures into alkaline cement, with pre-zeolite as secondary products (Fernández-  
119 Jiménez and Palomo, 2005; Fernández-Jiménez et al., 2006). The precursor should always be  
120 submitted to a previous thermal treatment, capable of inducing the loss of constituent water  
121 and the subsequent re-coordination of the aluminium and oxygen ions, transforming an  
122 originally crystalline structure into an amorphous one, more susceptible to further chemical  
123 reactions. The relative presence of calcium in the precursor and/or in the activator is very  
124 important, since the speed of the reactions is highly dependent on the type of aluminosilicate  
125 gel being formed, either N-A-S-H or C-S-H. The former needs longer periods in order to  
126 mature into a stable and reliable matrix, while the latter has curing / developing periods  
127 similar to those obtained with cement-based binders (Dombrowski et al, 2007; Garcia-Lodeiro  
128 et al., 2013).

129 Alkaline activation has recently started to be applied in soil stabilisation applications,  
130 using fly ash as the precursor, allowing significant strength gains (Zhang et al, 2013; Sukmak  
131 et al., 2013, 2015; Cristelo et al., 2011, 2012a, 2013; Silva et al., 2013; Rao and Acharya,

132 2014; Yi et al., 2015; Rios et al., 2015; Phummiphan et al., 2016), but also financial and  
133 environmental benefits, which proved very competitive when compared with lime or cement-  
134 based soil stabilizers (Cristelo et al., 2015). Other works were also recently published using  
135 alkaline activation to stabilise other types of granular materials that do not include soils such  
136 as construction and demolition waste (Mohammadinia et al., 2015; Arulrajah et al., 2016),  
137 coffee grounds (Kua et al., 2016), or recycled asphalt pavement (Saride et al., 2016, Hoy et  
138 al., 2016). However, constraints related with short deadlines regarding the entry into operation  
139 of most infrastructures, and namely road and railways, often require tight construction  
140 periods, which hinders the use of alkali activated low calcium (class F) fly ash, based on the  
141 mentioned slower reaction kinetics of the N-A-S-H gel, when compared with the C-S-H gel.  
142 However, in the presence of enough calcium, the two systems are very compatible, and  
143 several studies have focused on the characterisation of their interaction and coexistence  
144 (Garcia-Lodeiro et al., 2009, 2011; Puligilla & Mondal, 2013; Bui et al, 2015).

145 The need to improve the reaction kinetics of alkali activated class F fly ash binders  
146 constituted the motivation of the present paper. A viable method would have been the use of  
147 activated class C (high calcium) fly ash, which has proved very efficient for the stabilisation  
148 of a lateritic soil for bound pavement application (Phummiphan et al., 2016). However,  
149 assuming that such high calcium fly ash is not available, the need to develop alternative  
150 procedures to increase the strength gain rate is obvious. Therefore, several mixtures were  
151 prepared to find an effective method to increase the strength gain rate of a silty-sand stabilised  
152 with low calcium, alkali activated fly ash.

153 Some authors (Ghosh and Subbarao, 2001; Kumar et al., 2007; Samaras et al., 2008;  
154 Consoli et al., 2011b) have reported that a significant strength increase of lime stabilised  
155 clayey soils can be achieved with the addition of fly ash, enhancing the pozzolanic reaction,  
156 which is the main responsible for the mechanical properties of the soil-lime mixture.  
157 Additionally, several authors (Ramesh et al., 1999; Narendra et al., 2003; Cristelo et al., 2009)

158 have concluded that the formation of such pozzolanic reaction compounds is faster in the  
159 presence of sodium chloride. According to these authors, this can be attributed to the  
160 increased solubility of the silica and to the formation of sodium calcium silicate hydrate (N-S-  
161 C-H), more voluminous and with a higher water-holding capacity than the well-known C-S-H  
162 gel, formed in the sole presence of soil and lime. Based on these findings regarding the role of  
163 fly ash and sodium chloride on soil stabilisation, additional mixtures were considered, for  
164 comparison purposes, in which fly ash was added to soil-lime mixtures, with and without the  
165 addition of sodium chloride. The addition of sodium chloride was also used to prepare  
166 activated fly ash mixtures. Finally, a specific commercial additive for road pavement  
167 applications (recently available in the market called RoadCem®) was mixed with Portland  
168 cement, and the resulting binder was used to prepare some additional specimens. The result  
169 was regarded as a threshold value to which the mixtures developed in the present work were  
170 compared with.

171 In short, the following mixtures were conceptually analysed throughout the paper:

- 172 • Lime alone and with sodium chloride;  
173 • No-activated fly ash and lime with and without sodium chloride;  
174 • Activated ash and lime with and without sodium chloride;  
175 • RoadCem® and Portland cement

176

## 177 **Experimental program**

### 178 *Soil identification*

179 The work presented in this paper was based on a soil from Poland, which was fully  
180 characterized at the beginning of the experimental program described below. Typical  
181 geotechnical identification tests were performed in order to determine the particle size

182 distribution, Atterberg's limits, Modified Proctor compaction parameters, and the California  
183 Bearing Ratio (CBR). The particle size distribution curve (Figure 1) evidences a well graded  
184 soil, with almost 50% fines – silt (27.8%) and clay (16%), which was classified as a SC-SM -  
185 silty sand according to ASTM (2011) D 2487-11. Other parameters are summarised in **Erro!**

186 **A origem da referência não foi encontrada..**

187 The mineralogy was examined by a PANalytical X'Pert Pro diffractometer, fitted with  
188 an X'Celerator detector and secondary monochromator. The scans covered a  $2\Theta$  range of  $10^\circ$   
189 to  $80^\circ$ , with a nominal step size of  $0.017^\circ$  and 100 s/step. CuK  $\alpha$  radiation, with a wavelength  
190 of  $\lambda = 1.5418 \text{ \AA}$ , was used. Qualitative phase identification was made using High Score Plus  
191 software, which utilises the International Centre for Diffraction Data Power Diffraction File  
192 database (ICDD PDF-2, Sets 1-49, 1999) as a reference. X-Ray diffraction patterns of the soil  
193 showed the presence of quartz and muscovite on the mineralogical composition.

194 ***Characterisation of the binders***

195 The fly ash was provided by a Portuguese thermo-electric power plant and, based on ASTM C  
196 618 (2015), was classified as Class F due to the low calcium content as it is expressed on Table  
197 2. Hydrated lime provided by the company Lusical, Lhoist was also used. The particle size  
198 distribution of both lime and fly ash are plotted in Figure 1 together with the soil particle  
199 analysis. The activator solution used was composed by sodium hydroxide and sodium silicate  
200 . The former was supplied in pellets, with a specific gravity of 2.13 at  $20^\circ\text{C}$  (99 wt. %), as  
201 indicated by the supplier, which was then dissolved in water to pre-determined concentration  
202 of 12.5 molal. The sodium silicate technical sheet indicates a unit weight of  $1.464 \text{ g/cm}^3$  at  
203  $20^\circ\text{C}$ , a  $\text{SiO}_2/\text{Na}_2\text{O}$  weight ratio of 2.0 (molar oxide ratio of 2.063) and a  $\text{Na}_2\text{O}$  concentration  
204 in the solution of 13.0%. A sodium silicate to sodium hydroxide ratio of 0.5 was always  
205 considered. Deionised water was used in every mixture prepared during the work presented.

206 Some mixtures were also prepared with Ordinary Portland cement, type CEM I-42.5R,  
207 together with a commercial soil stabiliser, specifically designed for road pavements, named  
208 RoadCem®. This product can be described as a fine grain additive, based on alkali earth  
209 metals and synthetic zeolites, complemented with a complex activator. Based on the technical  
210 sheet, this product modifies and extends the chemistry of the cement hydration process and  
211 extends the crystallization process by forming long needle crystalline structures. It is able to  
212 delay or to speed up the hydration process of cement and can thus be used as a tool to custom  
213 design the mixes of required performance. It is always used in combination with cement  
214 and/or other pozzolanic materials (Marjanovic et al., 2009).

215 ***Specimen preparation and mixture composition***

216 Preparation of the soil included drying and de-flocculation by hand. The solids were then dry  
217 mixed for 10 min in a Hobart counter mixer, and the liquid phase was carefully added,  
218 requiring an additional 10 min mixing period. Cylindrical specimens with 70 mm in diameter  
219 and 140 mm in height were then compacted for uniaxial and triaxial compression strength  
220 tests, using static compaction according to EN 13286-53 CEN (2004a). The top and bottom of  
221 the moulds were then covered with cling film and the specimens were left for forty-eight  
222 hours before the mould could be removed. After demoulding the specimens were allowed to  
223 cure for longer in the same conditions: at  $20^{\circ}\text{C} \pm 1^{\circ}\text{C}$  and  $90\% \text{ RH} \pm 3\%$ .

224 Identification of the mixtures is presented in **Erro! A origem da referência não foi**  
225 **encontrada.** Each component of the solid phase is defined as a percentage of the total mass  
226 of solids, and the water content as the ratio between the amount of water and the amount of  
227 solids. All mixtures were identified using the following code: ‘S’ (soil); ‘FA’ (fly ash); ‘L’  
228 (lime); ‘C’ (sodium chloride); ‘AA’ (alkali activated); ‘OPC’ (Ordinary Portland cement) and  
229 ‘RC’ (RoadCem). Some of these letters were followed by a number, indicating the respective  
230 dosage. In the first step the specimens were moulded on the maximum dry unit weight and

231 optimum water content, obtained for the soil, using a modified Proctor test (**Erro! A origem**  
232 **da referência não foi encontrada.**), but in the second and third steps the presence of lime  
233 and the consequent cationic exchange and flocculation prevented a higher compaction degree.  
234 Therefore, in steps 2 and 3, mixtures were moulded at 88% of the optimum dry unit weight,  
235 corresponding to a maximum dry unit weight of 18.8 kN/m<sup>3</sup>. All specimens were moulded  
236 with a water content of 5%, corresponding to 88% value in the Proctor curve, with the  
237 exception of the geopolymeric mixtures, which required a higher water content in order to  
238 fulfil the criteria related with the activator/ash ratios. For comparison purposes, a specimen of  
239 unstabilised soil was also prepared, being compacted at 88% of the Proctor maximum value.

240 ***Uniaxial compression strength testing***

241 An Instron® electro-mechanical load frame , fitted with a 25 kN load cell, was used for the  
242 unconfined compressive strength tests. The tests were carried out according to EN 13286-41  
243 (CEN, 2003a) under monotonic displacement control, at a rate of 2 mm/s, and the entire  
244 stress-strain curve was obtained from each test.

245 ***Triaxial testing***

246 Consolidated undrained triaxial compression tests (CU), following CEN ISO/TS 17892-9  
247 (CEN, 2004b), were performed on the original soil, after isotropic effective consolidation  
248 stresses ( $\sigma'_c$ ) of 33 kPa, 100 kPa and 300 kPa. Additionally, drained triaxial compression tests  
249 (CD) were performed in two of the bounded mixtures – S\_FA15\_L5\_AA3 and  
250 S\_FA10\_L5\_AA3 – previously cured for 28 days, using effective isotropic consolidation  
251 stresses of 50 kPa, 100 kPa and 200 kPa. These particular mixtures were chosen due to the  
252 fact that they reached the most performing behaviour in the UCS tests.

253 After the installation in the triaxial cell, water was allowed to percolate through the  
254 specimen until a volume of water higher than twice the volume of voids was obtained. This

255 facilitated saturation because removed the air bubbles present in the specimen and in the  
256 tubing system. Then, the saturation followed where cell and back pressures were increased  
257 simultaneously at a rate of 30 kPa per hour keeping an effective stress of 10 kPa, until cell  
258 and back pressure achieved 410 and 400 kPa respectively. This back-pressure value was kept  
259 constant during 24 h, assuring full-saturation which was verified by the B Skempton  
260 parameter which was close to 1.

261 The same servo-hydraulic load frame used for UCS tests, fitted with a 25 kN load cell,  
262 was used to apply the deviatoric load, under monotonic displacement control, at a rate of  
263 0.01 mm/min. The axial deformation was measured by two Linear Displacement  
264 Transformers (LDTs) (Goto et al., 1991), while an additional LDT was installed to monitor  
265 the radial deformation as illustrated in Figure 2.

266 ***Seismic waves***

267 Compression seismic (P) wave velocities were measured by ultrasonic non-destructive  
268 transducers, after curing periods of 12, 24 and 48 h, and then after 3, 7, 14 and 28 days in  
269 some of the specimens tested in unconfined compression as expressed in **Erro! A origem da**  
270 **referência não foi encontrada.** A commercially available equipment was used, consisting of  
271 axially aligned piezoelectric transducers, with a nominal frequency of 54 kHz and 30 mm in  
272 diameter; a waveform generator; an amplifier and an interval timer with a direct reading  
273 digital display, which showed the P-wave velocity ( $V_p$ ) in real time. An oscilloscope would  
274 have been preferable, due to the lower definition of the pulse wave when travelling through  
275 heterogeneous materials, like soil. However, such equipment was not available. The  
276 equipment has an excitation voltage of 1000V and a resolution of 0.1 $\mu$ s. The equipment  
277 identified automatically the P wave travel time without any operator interference, and  
278 therefore no specific analysis was needed as in other advanced devices (Camacho-Tauta et al.,  
279 2015). In this case, standard EN 12504-4 (CEN, 2003b) was used as general reference, and

280 calibration of the transducers was automatically performed by the equipment, using a metallic  
281 rod with a known P-wave travelling time of  $20.9\mu\text{s}$ . The measurements were taken along the  
282 longitudinal axis of the specimens, with the specimen vertically aligned and the transducers  
283 installed on opposite faces. The acoustic coupling between the transducers and the specimen  
284 was obtained using ultrasound gel, while firmly pressing the transducers against the top  
285 surfaces of the specimen. Each result presented is the average of ten consecutive readings.

286

287 **Experimental results**

288 ***Unconfined compressive strength***

289 In the first step of **Erro! A origem da referência não foi encontrada.**, designed to  
290 obtain a guideline in terms of time needed for the reaction to produce meaningful UCS  
291 results, three different liquid/solids ratios were considered, by varying the liquid phase (the  
292 solids were kept at 85% S + 15% FA). Based on the results obtained, further activated  
293 mixtures were performed with an activator / ash ratio equal to 0.707 (the criteria for such  
294 value are discussed later in the text).

295 In the second step, the soil was mixed with different binders in order to evaluate the  
296 effect, on the reaction's kinetics, of each of the combinations proposed. Mixtures of soil-lime  
297 as well as soil-lime-fly ash (both with and without sodium chloride) were first performed with  
298 deionised water as the only constituent of the liquid phase. Then, the fly ash and fly ash-lime  
299 mixtures were prepared with the activator as the only constituent of the liquid phase.

300 A third and final step was then designed to reduce the mixture cost which included the  
301 reduction of the fly ash to 10% of the total solids (and consequently, a lower quantity of  
302 activator solution), while using only one lime percentage (10%) and 1% of sodium chloride.  
303 Also included in this final step were the OPC + RC mixtures.

304 At least 3 specimens of each condition were always moulded and tested so that the  
305 presented data corresponds to the average of three tests, improving the statistical confidence  
306 on the results. A unique curing period of 28 days was considered during the first step, while  
307 three different curing periods of 3, 7 and 28 days were considered in the second step. In the  
308 third step, L + FA specimens were tested after 7 days, while OPC + RC specimens were cured  
309 for 3, 7 and 28 days as expressed in **Erro! A origem da referência não foi encontrada.**

310 The results obtained in step 1 with distinct activator contents and a ash content of 15% was  
311 aimed to evaluate the effect of the Na<sub>2</sub>O/ash ratio on UCS (Cristelo et al., 2012a). As such,  
312 activator/ash ratios of 0.943, 0.825 and 0.707 were considered, resulting in activator contents  
313 (liquid/solids ratio) of 14.1%, 12.4%, and 10.6%. Since the three mixtures presented a  
314 significant strength increase (Figure 3), the lowest activator content of 10.6% was selected for  
315 the following steps, even though the activator content of 12.4% achieved higher strength. This  
316 was due to the fact that a reduction in activator implies a reduction of the total cost of the  
317 mixture, since the activator is, by a significant margin, the most expensive component.

318 Figure 4 illustrates the strength results obtained in step 2, as well as the corresponding secant  
319 stiffness modulus at 50% of the maximum strength ( $E_{50}$ ). This parameter is essential to  
320 characterise the material's constitutive model included in any finite element code, and  
321 therefore its calculation is extremely useful for design purposes. With a few exceptions, the  
322 mixtures with lower and higher strength values presented also lower and higher stiffness  
323 values, respectively. Mohammadinia et al. (2014) have reported a  $E_{50}$ /UCS ratio around 1000  
324 for cement-treated construction and demolition materials but in this case the ratio is closer to  
325 100 probably due to a finer grading of the soil.

326 Note the significantly higher strength, after every curing period, achieved by the  
327 mixtures with both lime and activated ash. While the mixture with activated ash achieved  
328 similar strength values to those obtained by the lime and lime-fly ash mixtures, the addition of

329 lime to the former resulted in a 4 times strength increase. This is a consequence of the faster  
330 reactions in C-S-H systems, relatively to N-A-S-H systems, as observed by the authors  
331 Cristelo et al. (2012b) during a previous work that compared the strength gain rate motivated  
332 by the activation of class C and class F fly ash.

333 Considering the existing design guidelines for roads, both the Portuguese road  
334 specifications (JAE, 1995) and the French (GTS, 2000) classification chart for soil-cement  
335 mixtures, suggest a minimum indirect tensile strength of 0.2 or 0.3 MPa, depending whether  
336 in situ or plant mixing is used. These values can be converted into UCS values of 2 and  
337 3 MPa assuming a ITS/UCS=10%. The GTS (2000) specification also adds the need to have  
338 1 MPa at 7 days of curing. Using these values as target for design, as expressed in Figure 4, it  
339 becomes clear that the soil-lime mixture do not fulfil these aims, while the soil-lime-ash  
340 mixtures stay slightly above these limits. For higher performances, corresponding to higher  
341 class treated layers as expressed in GTS (2000), activated ash mixtures are needed.

342 A final third step was conducted to address the possible reduction of the economic and  
343 financial cost of the binders, considering that most UCS values obtained were very  
344 satisfactory and could be slightly reduced without compromising its mechanical performance.  
345 This cost decrease was obtained with a reduction of the fly ash percentage to 10%, down from  
346 the step 2 overall value of 15%. Note that a reduction in fly ash content does not represent a  
347 significant economic or environmental saving, but the reduction of the alkaline solution  
348 content needed to activate the fly ash is, indeed, significant in terms of both of these aspects.  
349 Results presented in Figure 5 (after curing for 7 days) show that the mixtures with both lime  
350 and activated ash remain as the most performing combinations, even relatively to those  
351 mixtures based on both cement and RoadCem®.

352 ***Triaxial compression strength***

353 Figures 6 and 7 show the stress-strain curves obtained from the triaxial tests performed on the

354 soil and selected bounded mixtures (with lime and activated ash), respectively as indicated in  
355 **Erro! A origem da referência não foi encontrada.** These soil-binder mixtures were  
356 selected based on the efficiency/cost ratio obtained from the results from the uniaxial  
357 compression tests presented above.

358 The soil presented a quasi-linear behaviour up to 29%, 27% or 14% of the maximum  
359 deviatoric stress, depending on the effective confining pressure applied of 33, 100 or 300 kPa,  
360 respectively. After this linear segment, the soil stiffness significantly decreases, due to  
361 yielding, reaching a zero value when the soil achieved a constant stress plateau at the 20% to  
362 30% strain mark. **Erro! A origem da referência não foi encontrada.** presents the results of  
363 the stiffness modulus at 50% of the maximum deviatoric stress ( $E_{50}$ ), a usual design parameter  
364 as explained before.

365 The bounded mixtures showed completely different stress-strain curves, which in this  
366 case are typical of cemented materials (Rios et al., 2014, 2015), with a peak stress at very low  
367 strain levels (between 0.15 and 0.30%), followed by a very abrupt strain softening due to  
368 bond degradation. The material stiffness does not seem to be very much affected by the  
369 effective confining pressure, indicating that, for this pressure range, the consolidation stress  
370 did not produce an increase in stiffness, but it did not damage the cemented structure either.  
371 This behaviour, typical of cemented soils (e.g., Fernandez and Santamarina, 2001), is  
372 expressed in **Erro! A origem da referência não foi encontrada.** showing that the  $E_{50}$   
373 stiffness modulus was very similar for the three tests on S\_FA10\_L5\_AA3 specimens  
374 (approximately 3000 MPa) and also for the three tests on S\_FA15\_L5\_AA3 specimens  
375 (approximately 4000 MPa). Note that the latter  $E_{50}$  value is almost 10 times higher than the  
376 values obtained with the unstabilized soil. Also very clear is the fact that the reduction in ash  
377 content (from 15% to 10%), had a strong influence on peak stress, which decreased between  
378 30% to 40%, depending on the confining pressure.

379 **P-wave measurements**

380 The results obtained from the seismic wave measurements are presented in Figure 8 relating  
381 the P wave velocity ( $V_P$ ) with curing time. It should be noted that the specimens are not  
382 saturated, and so the presented results do not represent the propagation velocity through the  
383 water, but through the soil grains. Consequently, the positive evolution of  $V_P$  with curing time  
384 is expected since  $V_P$  are directly related with the constrained modulus being therefore an  
385 indirect measurement of stiffness. The relative position between the  $V_P$  values obtained in  
386 each specimen is in agreement with the previous data presented in Figure 4, confirming the  
387 sensibility of this measurement.

388 However, the major advantage of this data is that being a non-destructive method it is possible  
389 to have several measurements through the curing period without needing further specimens.  
390 Figure 8 shows that there is a significant improvement in the first 7 days of curing and then the  
391 evolution continues at a smaller rate. This is especially relevant for alkali-activated ash mixtures  
392 and for specimens with lime and sodium chloride.

393

394 **Discussion**

395 **Strength envelope of bounded and unbounded mixtures**

396 The strength increase resulting from the mixture with lime and activated ash can be quantified  
397 in terms of the Mohr Coulomb failure criterion, namely from the resulting strength parameters  
398  $\phi'$  (friction angle) and  $c'$  (cohesion). The peak and ultimate strength envelopes of the soil and  
399 the selected bounded mixtures are presented in Figure 9, while the corresponding Mohr-  
400 Coulomb parameters are summarized in **Erro! A origem da referência não foi encontrada..**  
401 It is possible to observe that the peak friction angle almost doubled, while the peak cohesion  
402 intercept suffered also a significant increase, which is typical of cemented materials. At

403 ultimate states the strength of the bounded mixtures approaches that of the soil since the  
404 cohesion intercept reduces drastically. However, the values of the ultimate state friction angle  
405 are still significantly higher than those obtained for the unstabilized soil (peak or ultimate  
406 state). This indicates that not all the artificially created bonds could be destroyed in the test,  
407 instead forming small clusters that reduced the void volume between the soil particles.

408 ***Effect of sodium chloride addition***

409 Figure 10 presents the Step 2 UCS related to the sodium chloride addition. The sodium  
410 chloride was particularly effective in increasing the strength development rate of the non-  
411 activated ash mixtures (S\_L5 and S\_FA15\_L5), particularly after 28 days. The contribution of  
412 the sodium chloride to the reaction kinetics is clear and, after 28 days, the addition of 1%  
413 represents an UCS increase of 69% of the lime-based mixture and 59% of the lime and no-  
414 activated ash mixture. Regarding the activated ash mixtures (S\_FA15\_L5\_AA3), it appears  
415 that the sodium chloride did not make a noticeable impact on strength increase, and has even  
416 hindered the short-term (3 days) behaviour.

417 ***Effect of lime percentage***

418 Step 2 tests also allowed to observe the effect of hydrated lime on the reaction kinetics of the  
419 mixtures. Figure 11 compares the strength values obtained for the two lime percentages  
420 considered (5% and 10%), for each of the three curing periods. The results indicate that the  
421 increase in lime content affects the rate of strength development, especially between the 7th  
422 and the 28th day. An increase of 5% in lime content, after 28 days curing, represented  
423 strength increases of 40% (soil + lime + water), 56% (soil + ash + lime + water) and 22%  
424 (soil + ash + lime + activator).

425    ***Effect of ash percentage***

426    Comparing step 2 and step 3 UCS results, after 7 days curing, it is possible to evaluate the  
427    effect of the fly ash content – 0%, 5% and 10% of the total solids weight (Figure 12). Overall,  
428    the results are in agreement with those from the triaxial tests presented before, where it was  
429    clear that the reduction in ash content (from 15% to 10%) represents a decrease in strength.  
430    This decrease is significantly more pronounced when the ash is activated, at least after 7 days  
431    curing.

432              On the other hand, it is interesting to note that the use of lime and 10% fly ash  
433    represents a strength decrease relatively to the use of lime alone. This is contrary to the results  
434    of Consoli et al. (2011b), that reported a significant strength increase when 12,5% and 25%  
435    ash were added to distinct lime percentages (from 3% to 9%), when stabilizing a similar soil,  
436    using very similar fly ash, in terms of chemical composition. This suggests the possibility of  
437    existing a minimum ash content in order to take full advantage of the pozzolanic effect of the  
438    ash. On the other hand, pozzolanic effects on strength require an extended curing period to be  
439    fully noticeable. It is possible that, for higher curing periods (when there is enough time to  
440    take full advantage of the added fly ash), the strength of the soil + lime + ash + water  
441    mixtures will increase linearly with the ash content. This is not noticeable at this early curing  
442    period of 7 days, where the overall particle size distribution might be prevalent in terms of  
443    UCS. Assuming that this second hypothesis is indeed correct, the gap between activated and  
444    non-activated mixtures will probably be reduced with time. Although both mixtures have ash  
445    and calcium required to develop pozzolanic reactions, in the case of the activated mixtures  
446    there is also a N-A-S-H gel type forming, which will hinder the development of calcium-  
447    based reactions after medium to long term curing periods.

448

449 ***Comparison of different binders***

450 Figure 13 compares the different binders considered: lime; activated ash; lime  
451 and activated ash; and OPC with RoadCem. As it was already observed in previous figures,  
452 the activated ash mixed with lime always showed the best performance, both in terms of final  
453 (28 days) strength and in terms of strength gain rate. Somehow surprisingly, the OPC +  
454 RoadCem mixture showed a rather slow strength increase, when compared with the remaining  
455 mixtures. However, it should be taken in consideration that the relative dosage might have a  
456 strong impact on these results, i.e., to have comparable strength values a higher dosage of  
457 OPC and RoadCem® might be necessary. On the other hand, the OPC content was already  
458 significant (8% and 12%), indicating that for the same strength values the use of alkali-  
459 activated ash may be more economic.

460

461 **Conclusions**

462 The paper compares the performance of different binders, based mainly on lime and fly ash,  
463 with or without alkaline activation as the liquid phase. It is highlighted that the alkaline  
464 activation reaction kinetics may be improved or reduced depending on the mixture  
465 composition. Lime, ash, sodium chloride and the alkaline activator were used in different  
466 combinations and compared to Portland cement and a specific new additive, recently available  
467 in the market for road applications. The following conclusions may be taken:

- 468     • The lime + activated ash mixtures were the most performing in terms of uniaxial  
469         compression strength and axial stiffness;
- 470     • The sodium chloride significantly improved the lime and lime-ash mixtures, but  
471         provided only a slight improvement in the activated ash mixtures;
- 472     • A 5% increase (from 10% to 15% in terms of solids weight) in lime content  
473         represented an important strength increase after 28 days, for the lime-based mixtures,

474 for the lime and no activated ash mixtures and for the lime and activated ash mixtures.

475 Stress-strain curves and strength envelopes obtained in triaxial tests for the best

476 performing mixtures (with lime and activated ash) showed a typical strong cemented

477 soil behaviour with friction angles above 50° and cohesion intercepts higher than

478 500kPa.

- 479 • Considering the existing design guidelines for roads, the soil-lime mixtures studied in  
480 this work did not fulfil the necessary requirements, while the soil-lime-ash mixtures  
481 stay above the minimum values even without the activator. For higher performances,  
482 corresponding to higher-class treated layers as expressed in GTS (2000), activated ash  
483 mixtures are needed.

484

485 From the results obtained it is possible to conclude that alkali activated binders, based on low-  
486 calcium fly ash, can be very competitive with cement, regarding soil stabilisation for transport  
487 infrastructures applications. If the calcium content of the original fly ash is increased using an  
488 additional component like lime, not only the strength gain rate is enhanced, but the final  
489 strength is also improved.

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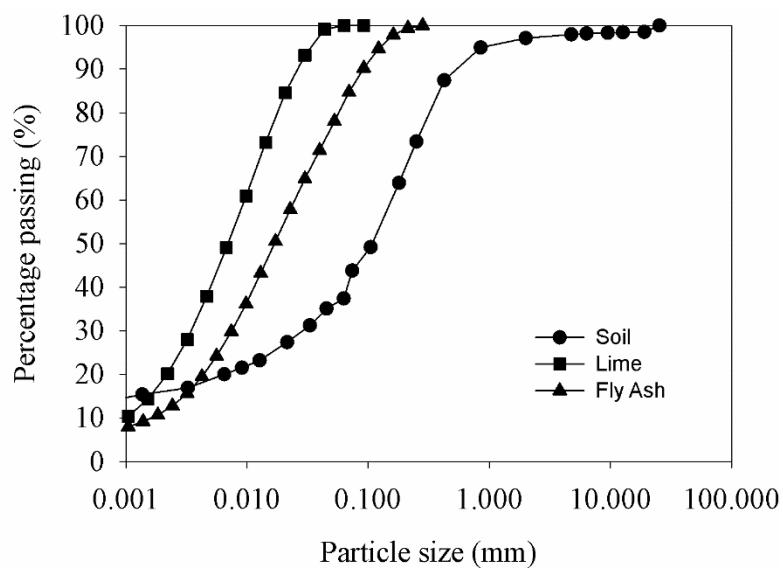
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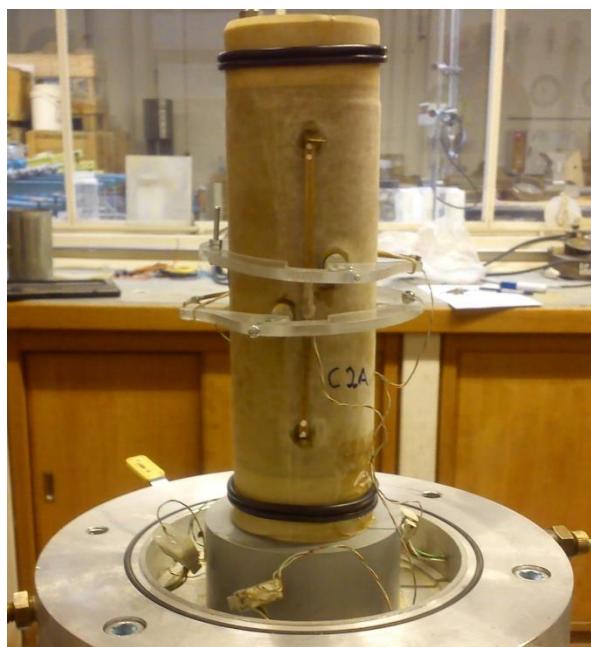
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734 **List of Figures**



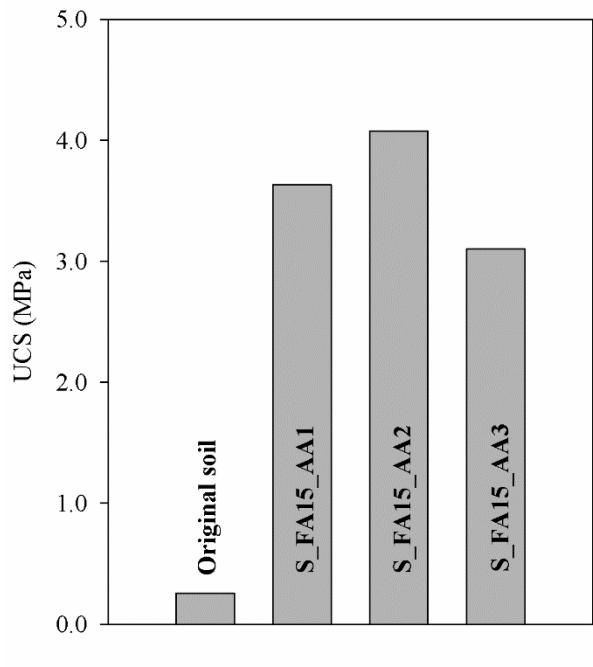
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736 Figure 1 - Particle size distribution of the soil, lime and fly ash



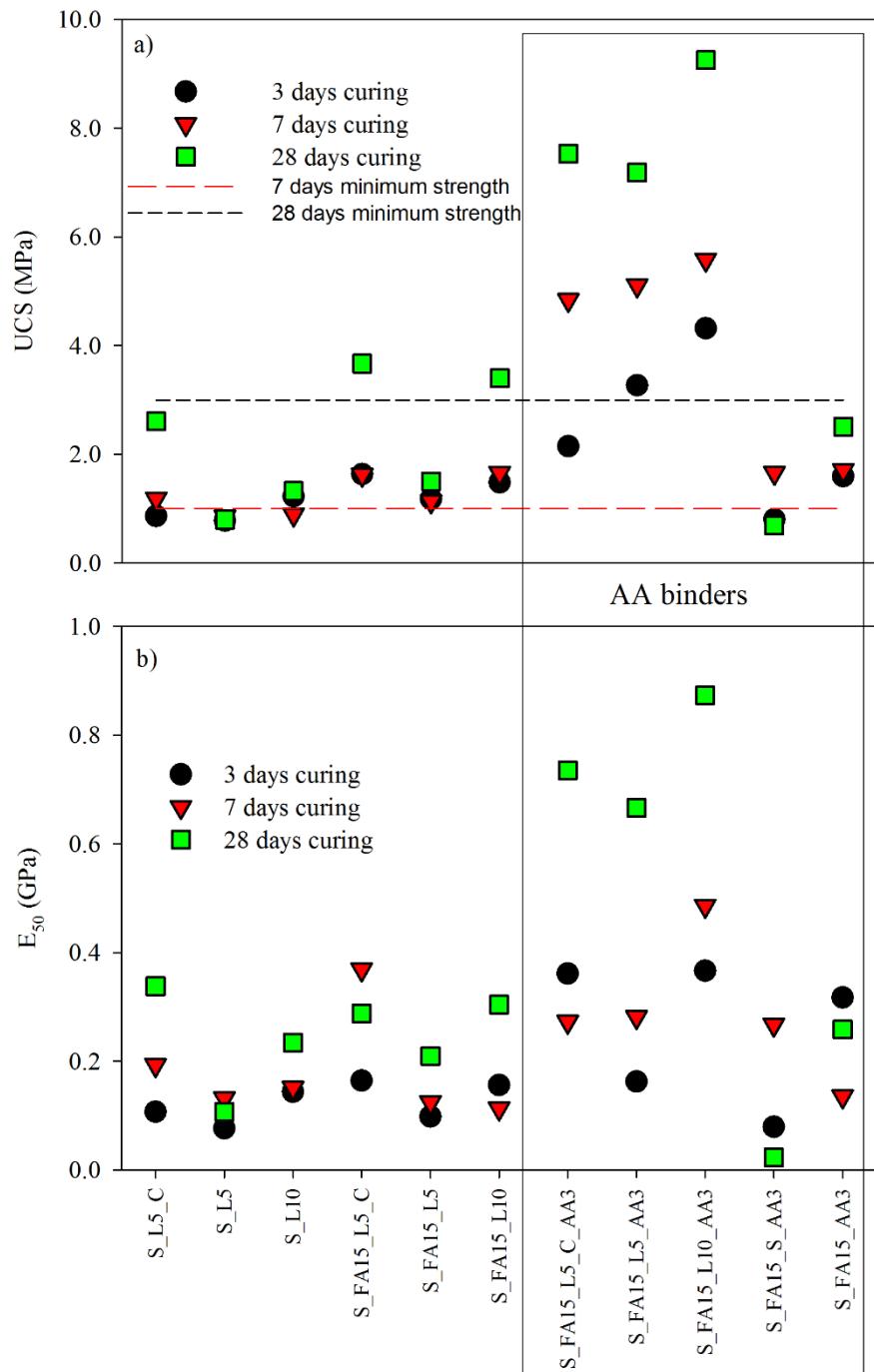
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738 Figure 2 - Triaxial test setup showing the Linear Displacement Transformers (LDT) sensors



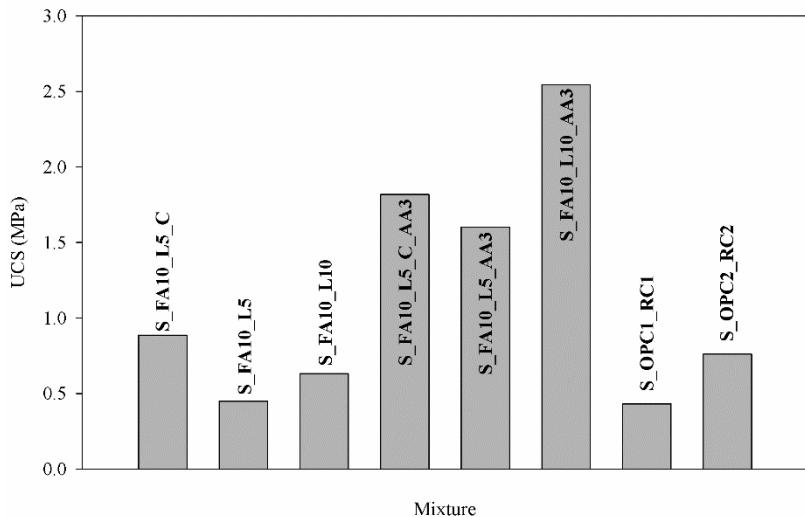
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740 Figure 3 - Step 1 UCS results (28 days curing)



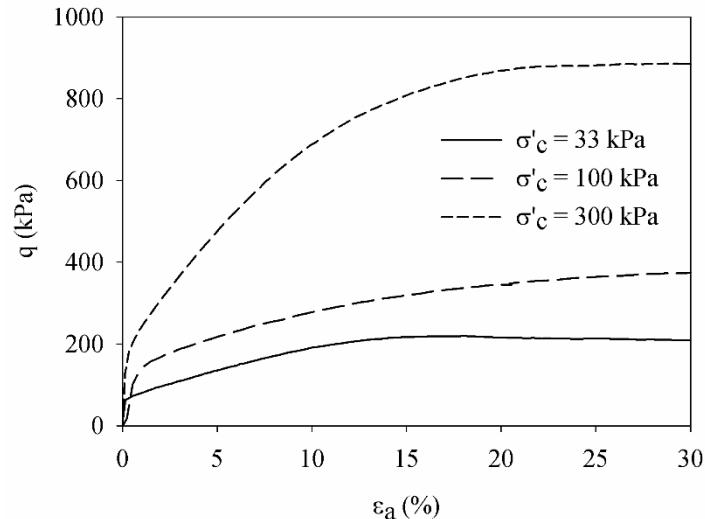
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742 Figure 4 - UCS results for step 2 (3, 7 and 28 days curing): a) unconfined compressive  
 743 strength (UCS); b) Stiffness modulus at 50% of maximum strength ( $E_{50}$ ) - S (soil); FA (fly  
 744 ash); L (lime); C (sodium chloride); AA (alkali activated)



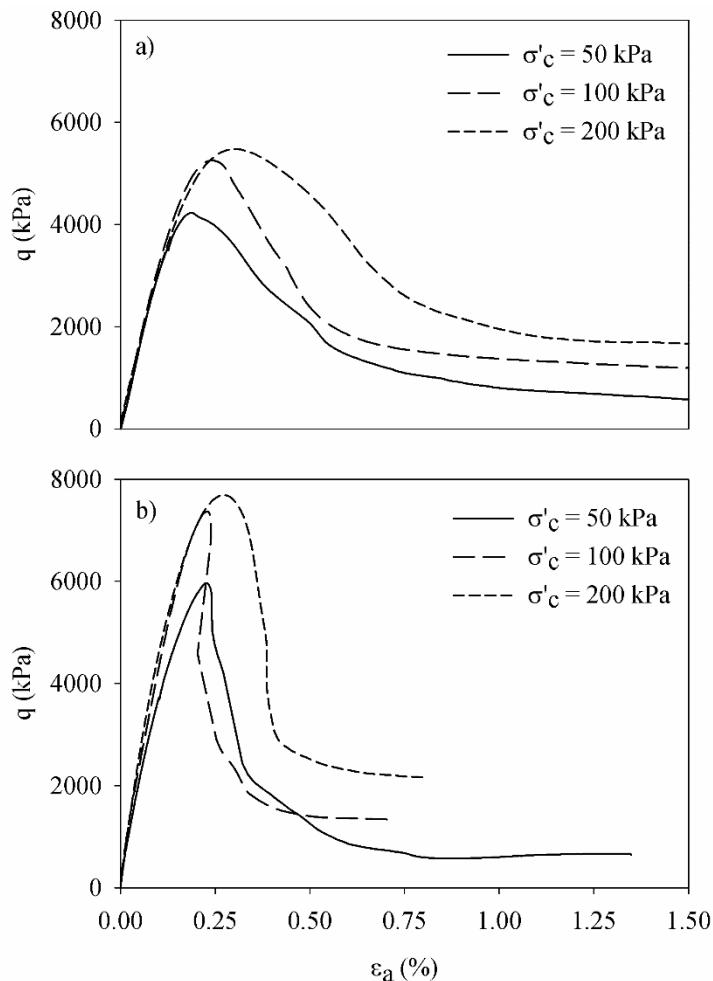
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746 Figure 5 - UCS results for step 3 (7 days curing) - S (soil); FA (fly ash); L (lime); C (sodium  
 747 chloride); AA (alkali activated); OPC (Portland cement); RC (RoadCem)



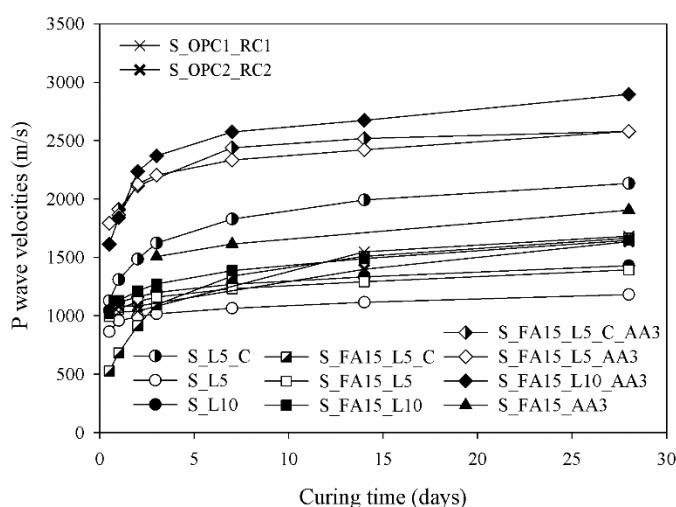
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749 Figure 6 - Stress-strain curves obtained from the triaxial compression tests performed on the  
 750 original soil ( $\sigma'c$  refers to the isotropic consolidation stress applied)



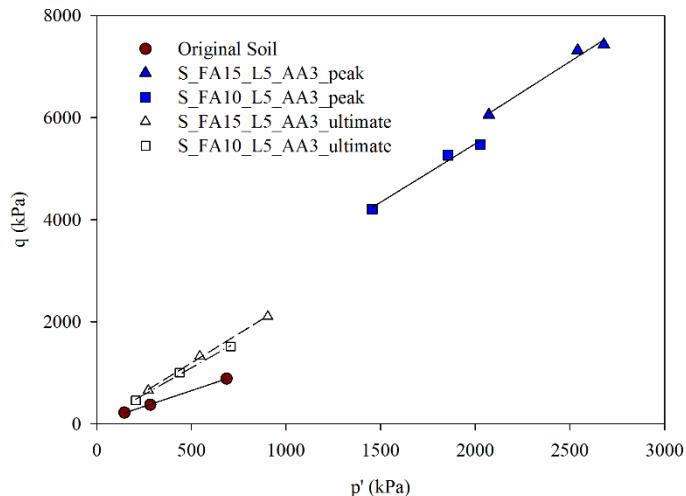
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752 Figure 7 - Stress-strain curves obtained from the triaxial compression tests performed on the  
 753 mixtures S\_FA10\_L5\_AA3 (a) and S\_FA15\_L5\_AA3 (b) ( $\sigma'_c$  refers to the isotropic  
 754 consolidation stress applied)



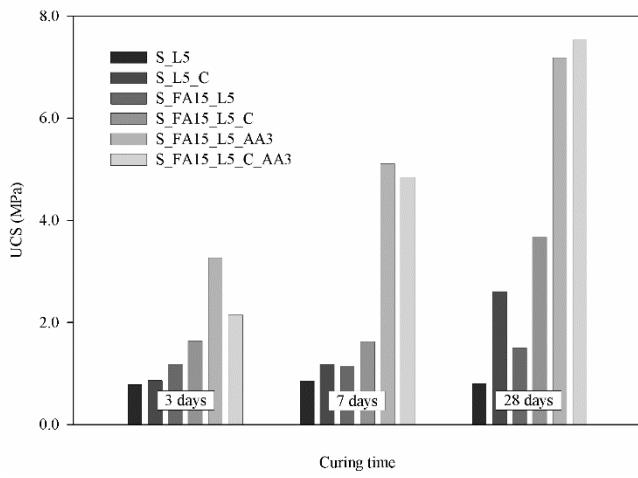
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756 Figure 8 – P wave velocities evolution with curing time



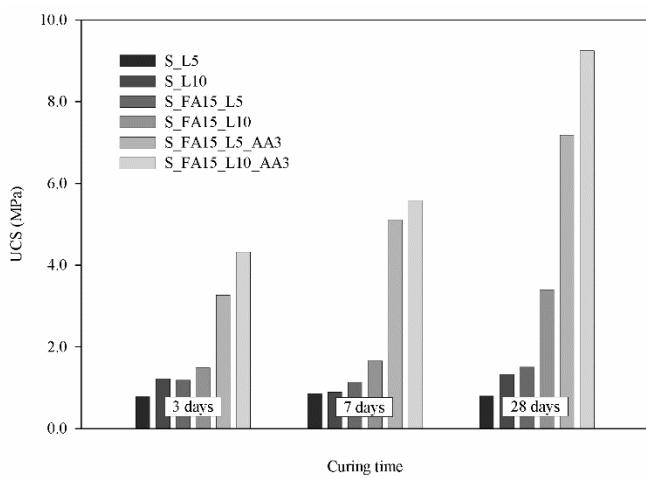
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758 Figure 9 - Peak and ultimate state strength envelopes (the soil specimens presented very  
759 similar peak and ultimate envelops, which are not possible to distinguish at this scale)



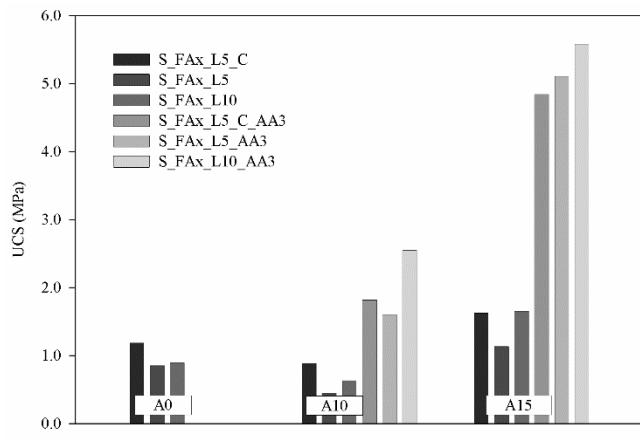
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761 Figure 10 - Effect of sodium chloride on the uniaxial compression strength



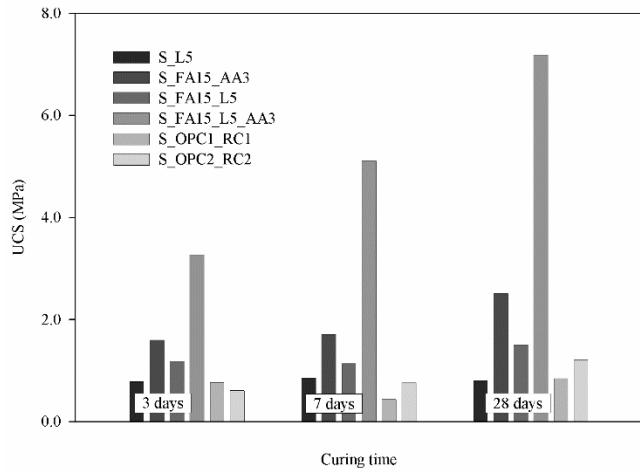
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763 Figure 11 - Effect of lime percentage on the uniaxial compression strength



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765 Figure 12 - Effect of ash percentage on the uniaxial compression strength after curing for 7  
766 days



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768 Figure 13 – Effect of the binder type on the uniaxial compression strength

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772 **List of Tables**

773

774 Table 1 - Main geotechnical properties of the soil

Properties	Symbol	Value
Plastic limit (%)	$w_p$	13.0
Liquid limit (%)	$w_L$	19.5
Average diameter (mm)	$D_{50}$	0.105
Fines fraction (sieve N° 200) (%)	$P_{#200}$	43.8
Uniformity coefficient	$C_u$	167
Curvature coefficient	$C_c$	6.7
Specific gravity	$G$	2.60
Modified Proctor test optimum water content (%)	$w_{op}$	8.2
Modified Proctor test maximum density (kN/m³)	$\gamma_d$	21.29
California Bearing Ratio (%)	CBR	9
Unified Soil Classification System ASTM D 2487 (2011)	-	SC-SM silty sand

775

776 Table 2: Chemical composition of the fly ash (wt%)

Element	Si	Al	Fe	Ca	K	Ti	Mg	Na	S	P
Fly ash	48.81	21.77	14.74	3.85	4.42	1.79	1.56	1.31	1.17	0.58

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Table 3 - Identification of prepared mixtures

Step	ID	Solid Phase				Liquid Phase	
		Soil (%)	Ash (%)	Lime (%)	SC(*) (%)	Activator/ Fly ash (wt)	Water content (%)
0	Soil	100	-	-	-	-	5.0
1	S_FA15_AA1	85	15	-	-	0.943	10.4
	S_FA15_AA2	85	15	-	-	0.825	9.1
	S_FA15_AA3	85	15	-	-	0.707	7.8
2	S_L5_C	94	-	5	1	-	5.0
	S_L5	95	-	5	-	-	5.0
	S_L10	90	-	10	-	-	5.0
	S_FA15_L5_C	79	15	5	1	-	5.0
	S_FA15_L5	80	15	5	-	-	5.0
	S_FA15_L10	75	15	10	-	-	5.0
	S_FA15_C_AA3	84	15	-	1	0.707	7.8
	S_FA15_AA3	85	15	-	-	0.707	7.8
	S_FA15_L5_C_AA3	79	15	5	1	0.707	7.8
	S_FA15_L5_AA3	80	15	5	-	0.707	7.8
	S_FA15_L10_AA3	75	15	10	-	0.707	7.8
	S_FA10_L5_C	84	10	5	1	-	5.0
3	S_FA10_L5	85	10	5	-	-	5.0
	S_FA10_L10	80	10	10	-	-	5.0
	S_FA10_L5_C_AA3	84	10	5	1	1.06	7.8
	S_FA10_L5_AA3	85	10	5	-	1.06	7.8
	S_FA10_L10_AA3	80	10	10	-	1.06	7.8
	S_OPCL_RC1	92	8% OPC + 0.08% RC			-	5.0
	S_OPCL_RC2	88	12% OPC + 0.11% RC			-	5.0

778 (\*) SC stands for sodium chloride

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Table 4 – Tests performed in each mixture and corresponding curing time

Step	ID	Curing time of the specimens tested in:		
		UCS	Triaxial tests	Seismic waves
0	Soil	0	0	
1	S_FA15_AA1	28		
	S_FA15_AA2	28		
	S_FA15_AA3	28		
2	S_L5_C	3, 7, 28		12, 24 and 48 h + 3, 7, 14 and 28 days
	S_L5	3, 7, 28		12, 24 and 48 h + 3, 7, 14 and 28 days
	S_L10	3, 7, 28		12, 24 and 48 h + 3, 7, 14 and 28 days
	S_FA15_L5_C	3, 7, 28		12, 24 and 48 h + 3, 7, 14 and 28 days
	S_FA15_L5	3, 7, 28		12, 24 and 48 h + 3, 7, 14 and 28 days
	S_FA15_L10	3, 7, 28		12, 24 and 48 h + 3, 7, 14 and 28 days
	S_FA15_C_AA3	3, 7, 28		
	S_FA15_AA3	3, 7		12, 24 and 48 h + 3, 7, 14 and 28 days
	S_FA15_L5_C_AA3	3, 7, 28		12, 24 and 48 h + 3, 7, 14 and 28 days
	S_FA15_L5_AA3	3, 7, 28	28	12, 24 and 48 h + 3, 7, 14 and 28 days
	S_FA15_L10_AA3	3, 7, 28		12, 24 and 48 h + 3, 7, 14 and 28 days
3	S_FA10_L5_C	7		
	S_FA10_L5	7		
	S_FA10_L10	7		
	S_FA10_L5_C_AA3	7		
	S_FA10_L5_AA3	7	28	
	S_FA10_L10_AA3	7		
	S_OPCL_RC1	3, 7, 28		12, 24 and 48 h + 3, 7, 14 and 28 days
	S_OPCL_RC2	3, 7, 28		12, 24 and 48 h + 3, 7, 14 and 28 days

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Table 5 - Stiffness modulus at 50% of the deviatoric stress ( $E_{50}$ ). Results in MPa

Mixture	Confining pressure of 33 (soil) or 50 kPa (mixtures)	Confining pressure of 100 kPa	Confining pressure of 300 (soil) or 200 kPa (mixtures)
Soil	3.68	6.84	10.21
S_FA10_L5_AA3	3194.3	3299.7	3154.3
S_FA15_L5_AA3	3901.1	4392.1	4759.9

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Table 6 – Mohr Coulomb failure criteria parameters

Mixture	Peak $\phi' (\circ)$	$c'$ (kPa)	Ultimate strength	
			$\phi' (\circ)$	$c'$ (kPa)
Soil	31	15	32	0
S_FA10_L5_AA3	56	587	49	27
S_FA15_L5_AA3	58	796	56	33

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